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Hydrogen Energy and its storage Overview on the activity of the Laboratory of Applied Mechanics in this field of research



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Context

A worrying situation???

- pollution by emission of green-house effect gas
- a scheduled collapse of fossil fuel energy
- ... renewable energy carriers





2005 Hydrogen Market, Hydrogen R&D and Commercial Implication in The U.S. and E.U. Fuji-Keizai USA, Inc., May 2005

and hydrogen is a promising one!!

Hydrogen set of problems



Hydrogen storage carriers



Gaseous... high pressure

Materials-based hydrogen storage

Absorption: metal hydrides



Adsorption : active coal, carbon nanotubes



Chemical reaction:

Hydrolysis : NaBH₄ + 2 H₂O <--> NaBO₂ + 2 H₂

(Sodiumhydridoborate)

Hydrogenation : C10H18 <--> C10H8 + 5 H2

(Dimethyl Propylmethane)



Materials-based hydrogen storage

Advantages : - safety (low pressure and temperature)

- stability

Drawbacks : - cost (rare components, catalyst)

- technological breakthrough
- weight (hydrides)
- off-board regeneration

Gaseous hydrogen storage

Method : gas is compressed 5000-10000 PSI



Advantages : - a well-known technology

- acceptable refueling time

Drawbacks : - low volumetric density

- HP requires specific equipements
- 10% heating power

Cryogenic hydrogen storage

Method : hydrogen is liquefied 20 K



Advantages : - a good volumetric density

- refueling technology is acquired

Drawbacks : - boil-off phenomenon (1%/day)

- Cryogeny requires specific equipements
- 30% heating power

Storage carriers comparison



Unreachable target ??!!

Hydrogen storage challenges

For transportation... 500 km requires 5 kg H₂

Weight and volume... to be reduced

Efficiency... to be increased

Durability... lifetime of 1500 cycles

<u>Refueling time</u>... less than three minutes

<u>Cost</u>... cheaper materials and compounds

<u>Codes and standards</u>... must be established

Analysis of the <u>full life-cycle</u> is missing

Optimization of a Type III hydrogen storage vessel (1)

Design of a hybrid storage tank (2)

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LMARC, Institut FEMTO ST P2M departement D. Chapelle, D. Perreux, F. Thiébaud...

What is a type III hydrogen storage vessel?



5 cm



What is the goal?

Development of a tank reinforced by composite

- in use pressure : 10000 PSI (70 Mpa)

- gravimetric density : 6%

- metal liner : stainless steel or aluminium alloy

- burst pressure : 23500-30000 PSI

The industrial approach

is based on time considerations

Polar winding

Circumferencial winding



The winding angle depends on the liner geometry

AXIAL LOADING



HOOP LOADING

Simple mechanical analysis

Optimal winding angle is 55°

Thin shell under internal pressure with close-end effect $\int \sigma_{zz}$



Strength equilibrium: $(\sigma_{\theta\theta} / \sigma_{zz}) = 2$ $\sigma_{\theta\theta} = Pressure$. Radius / thickness $\sigma_{zz} = Pressure$. Radius / (2 . thickness)



It means... $\tan^2 \alpha = 2$

First objective

Develop experimental techniques

- to get Type III pressure storage vessel
- to test these prototypes under internal pressures



Comparison with numerical tools

Manufacturing process (1)





Gauges

4

3

6

5

Some results



9 carbon layer composite on an aluminum liner Burst pressure: 780 bars Volume 0.8 litre





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1

Axial strain

Hoop strain



Second objective

Analytic modelling of the cylindrical section of a high pressure gas storage for thermomechanical loading



Numeric tool to design vessel structures combining a metal liner and a composite laminate (stiffness)

(Type III hydrogen storage tank)

Mechanical analysis (1)



Structure parameters

- liner inner radius R_0 , thickness e

- layers n_s , k^{th} layer thickness ep(k), the stacking sequence

Mechanical behaviour

- liner : elastic-plastic material (von Mises)
- composite : CLT, damageable, Tsai-Wu failure



Mechanical analysis (2)

Specific displacement field:

- U = U(r), V = V(r, Z), W = W(Z)
- constant axial strain along radial direction
- no dependency of $\varepsilon_{z\theta}$ with z coordinate

Boundaries conditions

- radial displacement and stress continuity
- internal pressure with close-end effect



Linear problem A.X = B













Design of a hydride storage tank Pr. Figiel, N.B. Selvaraj

$$\frac{\partial^2 \theta}{\partial r^2} + \frac{1}{r} \left(\frac{\partial \theta}{\partial r} \right) + \frac{g}{k} - \lambda \frac{\partial \theta}{\partial t} = 0$$

- $\begin{array}{lll} \Theta & \text{Temperature, }^{\circ}\text{C} \\ C_p & \text{Heat capacity, J kg}^{-1} \text{ K}^{-1} \\ h & \text{Heat transfer coefficient} \\ t & \text{Time sec} \end{array}$
- o Material density
- λ Thermal conductivity
- g Heat sources

Explicit resolution

 $\theta_{i,j+1} = \frac{\Delta t}{\lambda \Delta r^2} \left(\theta_{i+1,j} + \theta_{i-1,j} - 2\theta_{i,j} + \frac{\Delta r}{r} \left(\theta_{i+1,j} - \theta_{i,j} \right) + \Delta r^2 \frac{g}{k_1} \right) + \theta_{i,j}$

 $k = \frac{\rho C_P}{\lambda}$



From experimental data of absorption at 20, 50 and 80°C... LaNi_{4.78}Sn_{0.22} Heat sources are assumed under the form...

texp g b^2

 $LaNi_{4.78}Sn_{0.22} + nH_2 \rightarrow LaNi_{4.78}Sn_{0.22}H_{2n} + Q$





Profiles of heat produced during absorption. a) Temperature dependence, b) Temperature dependency of *b* coefficient.

Simulation of LaNi_{4.78}Sn_{0.22} absorption for a radius bed of 40 mm and a thickness tank of T mm a) Absorption rate versus time, b) Temperature profile versus time



T = 12 mm



Experimental investigations on various geometry of tank Temperature measurements on the external surface





Hydride tank Hydrogen bottle

Seivert apparatus in the LMARC

Conception d'un réservoir hybride de stockage intégrant les technologies de stockage solide et gazeux

Thesis Marko Feldic, HyTrain program CNR Florence, M. Zoppi CEA Grenoble, O. Gillia

Modélisation thermodynamique et cinétiques d'absorption d'hydrogène associées aux transformations de phase

Thesis Germain Gondor Pr. Christian Lexcellent

Original equipments

High pressure hydrogen equipment : 800 bars





Conclusions

- not only one energy carrier, not only one way to store hydrogen
- many improvements (almost on materials)
- solutions for future will depend on the investments

http://www.hydrogennow.org http://www.hydrogenus.com/ http://www.imr.salford.ac.uk/hytrain/ <u>http://www.eere.energy.gov/</u> http://www.storhy.net/ http://www.afh2.org http://www.h2eco.org/

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Thank you for your attention!!